

Synthesis and Morphological Reversion Activity on src^{ts}NRK Cells of Pyrimidinylpropanamide Antibiotics, Sparsomycin, Sparoxomycin A₁, A₂, and Their Analogues

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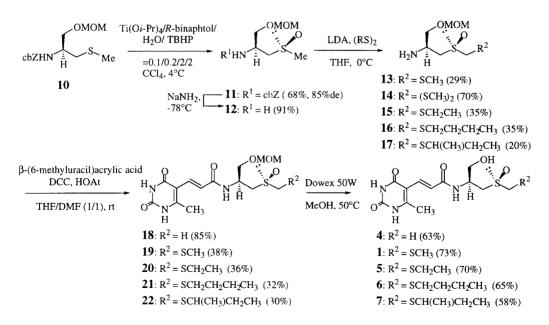
Abstracts: Three pyrimidinylpropanamide antibiotics sparsomycin (1), sparoxomycins A1, A2 (2, 3), and also six analogues (4-9) have been synthesized by employing asymmetric sulfide oxidation conditions as a key step. Sparsomycin (1) and its alkyl analogues (5-7) showed higher morphological reversion activities on srctsNRK cells than 2 and 3. © 1998 Elsevier Science Ltd. All rights reserved.

Sparsomycin (1), an inhibitor of protein biosynthesis, is a metabolite of *Streptomyces sparsogenes* or *Streptomyces cuspidosporus*. ¹⁻³ The structure of 1 is closely related to that of sparoxomycins A1 (2) and A2 (3), which were isolated from a culture broth of *Streptomyces sparsogenes* SN-2325 as new members of the pyrimidinylpropanamide antibiotic in 1996, ^{4a} differing only in their oxidation level at sulfur atom. The sparoxomycins (2 and 3) converted transformed morphology of temperature-sensitive mutant Rous sarcoma virus-infected NRK cells (*src*^{tS}NRK cells) to normal morphology at a wide range of concentrations without cytotoxicity. ^{4b} The sparsomycin (1) has antitumor activity; however, there are no reports regarding normalization of the phenotype of oncogene-transformed cells by 1 and its analogues. ⁵ Therefore, we became interested in designing and synthesizing analogues of these pyrimidinylpropanamide antibiotics, to evaluate their biological properties. We wish to report here a novel and stereoselective synthesis of natural products (1-3), and their analogues (4-9) as well as their morphological reversion activity on *src*^{tS}NRK cells. ⁶

Figure 1. Structures of sparsomycin (1), sparoxomycins A_1 , A_2 (2, 3), and their analogues 4-9.

The synthetic pathway to 1-9 is based on the following procedure which improved a shortcoming in Helquest's protocol² developed in the synthesis of sparsomycin (1). In order to achieve the stereoselective synthesis, the oxidation of sulfide (10) to the chiral sulfoxide (11) under asymmetric oxidation conditions⁷⁻¹¹ was studied. After several attempts, 12 diastereoselective oxidation was found to proceed catalytically in the presence of a titanium complex produced from Ti(Oi-Pr)₄ and chiral binaphtol. When 0.2 mol equiv. of R-(+)binaphtol was used as a ligand at 4°C in CCl₄, S configuration of sulfoxide (11) was isolated in 68 % yield with 85 % diastereomeric excess (de). 13 Single recrystallization of the diastereomixture from hexane-CH₂Cl₂ afforded optically pure (S)-11, mp. 113-114 °C, $[\alpha]_D^{19} = +87.1$ (c 0.19, CH₂Cl₂). ¹⁴ S-Sulfoxide (11) was subjected to sulfenylation after removal of the benzyloxycarbonyl (cbZ) group under dissolved metal conditions (91%). The sulfenylation of 12 to dithioacetal mono-oxide (13) was difficult to reproduce even under the reported conditions.² Only the double sulfenylation product (14) was obtained in 70 % by the addition of dimethyl disulfide after treatment of amine (12) with 2 equiv. of lithium disopropylamide (LDA). When 1 equiv. of dimethyl disulfide was first added to a THF solution of 12 and then LDA (1.5 equiv.) was added in two portions (reverse addition), sulfenylation proceeded to give the desired dithioacetal mono-S-oxide (13) in 29 % (53 % based on consumed material) along with 11% of 14. The coupling of β -(6-methyluracil)acrylic acid and 13 was achieved by 1,3-dicyclohexylcarbodiimide (DCC) and 1-hydroxybenzo-7-azatriazole (HOAt) to give 19 in 38 %. Hydrolytic removal of the MOM group with Dowex 50W in MeOH afforded 1, mp 204-206°C, in 73 % yield. The synthetic material was spectroscopically (IR, ¹H-NMR, ¹³C-NMR, MS) identical with natural sparsomycin, and also had specific rotation, $\left[\alpha\right]_{0}^{25} = +65.1^{\circ}$ (c 0.28, H₂O), in good agreement with the literature value [lit. 2b [α]_D = +65.1 (c 0.37, H₂O)]. The alkyl (Et, nBu and isoBu) analogues (5-7) and methyl sulfoxide analogue (4) were also prepared by employing the sequences similar to that developed for the synthesis of 1 from 12. These results are summarized in **Scheme 1**.

Scheme 1. Synthesis of sparsomycin (1) and its analogues (4-7).



Scheme 2. Synthesis of sparoxomycin A1 (2) and A2 (3).

reagents and conditions: a) $NaIO_4$, $MeCN/H_2O=1/1$, 23/24=1/1, 77% b) Dowex 50W, MeOH, $50^{\circ}C$, 80% and separation by HPLC, 14% for 2, 12% for 3.

Scheme 3. Synthesis of cinnamide derivatives (8 and 9).

13
$$\xrightarrow{a \text{ or } b}$$
 \xrightarrow{R} \xrightarrow{N} \xrightarrow{H} \xrightarrow{H} \xrightarrow{H} \xrightarrow{S} \xrightarrow{S} \xrightarrow{C} \xrightarrow{S} \xrightarrow{C} \xrightarrow{R} \xrightarrow{H} \xrightarrow{H} \xrightarrow{H} \xrightarrow{S} \xrightarrow{S} \xrightarrow{C} \xrightarrow{S} \xrightarrow{S} \xrightarrow{C} \xrightarrow{R} \xrightarrow{H} \xrightarrow{H} \xrightarrow{H} \xrightarrow{S} \xrightarrow{S} \xrightarrow{C} \xrightarrow{S} \xrightarrow{S} \xrightarrow{C} \xrightarrow{H} \xrightarrow{H} \xrightarrow{H} \xrightarrow{H} \xrightarrow{H} \xrightarrow{S} \xrightarrow{S} \xrightarrow{S} \xrightarrow{C} \xrightarrow{S} \xrightarrow{S} \xrightarrow{S} \xrightarrow{C} \xrightarrow{S} \xrightarrow

reagents and conditions: a) trans-cinnamoyl chloride, Et₃N, DMAP, CH₂Cl₂, rt b) trans-3.4-dihydroxycinnamic acid DCC, HOAt, DMF, rt c) Dowex 50W, MeOH, 50°C

The 1:1 mixture of dioxodithioacetal **2** and **3** was prepared by NaIO₄ oxidation^{15,16} of **19** followed by removal of the MOM group in 62 % (2 steps). Sparoxomycins A1 (**2**) and A2 (**3**) were isolated by careful ODS column chromatography and HPLC (Mightysil RP-18) in 14 % for **2** and 12 % for **3**, respectively (**Scheme 2**).¹⁷ The cinnamide and catechol derivatives (**8** and **9**) were synthesized from **13**. The 65 % yield of **25** was obtained from coupling of (*E*)-cinnamoyl chloride with **13** under Et₃N, DMAP conditions. The analogue (**26**) was also obtained from coupling of 3,4-dihydroxycinnamic acid with **13** under DCC, HOAt conditions in 38 % yield. Hydrolytic removal of the MOM group of **25** and **26** with Dowex 50W in MeOH afforded **8** and **9** in 65% and 62 %, respectively (**Scheme 3**).

The morphological reversion activity on $snc^{tS}NRK$ cells (a gift from Dr. Y. Uehara, National Institute of Infectious Diseases) was assessed with synthetic compounds (1-9 and 19). ¹⁸ The results shown in **Table 1** disclose that the morphological reversion activity of 1 was found to be 6.5 μ M and this activity was 80 times more potent than that of 2 and 3. The alkyl analogues 5, 6 and 19, and 7 were 3-, 10-, and 30-fold less potent than 1. The methyl sulfoxide (4), cinnamide and catechol derivatives (8 and 9) did not show any activities at 530 μ M. These findings suggest that the pyrimidinylpropanamide group is essential and the monooxodithioacetal group plays an important role in exhibiting the morphological reversion activity.

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Compound	MEC (μM) ^a
sparsomycin (1)	6.5
sparoxomycin A1 (2)	530
sparoxomycin A2 (3)	530
MOM-1 (19)	60
4	no activity ^b
5	20
6	60
7	180
8	no activity ^b
9	no activity ^b

Table 1. Morphological reversion activity of srctsNRK cells

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- 12. We searched for the oxidation conditions under which the highest yield and selectivity could be obtained; thus, the combination of a protecting group on the hydroxy group, a ligand for catalyst (tartaric acid derivatives, binaphtols, salen and camphorsulfonyloxaziridine derivatives), the effects of the solvent (CCl₄, CH₂Cl₂, toluene), and the reaction temperature were each studied.
- The diastereomer excess was determined by HPLC analysis to be 85 % (Mightysil RP-18, MeOH/H₂O, 4/6, 1.0 mL/min) ^tR(S)-1 1, 14.2 (92.5 %); ^tR(R)-1 1, 15.3 (7.5 %).
- 14. The spectroscopic data (IR, ¹H-NMR) had in good agreement with the literature value. ²b Data for optically pure (S)-1 1; Rf: 0.15 (CH₂Cl₂/MeOH, 20/1); ¹H NMR (CDCl₃, 400MHz) 7.36-7.29 (5H, m, Ph), 5.77 (1H, br d, NH, J=7.5 Hz), 5.10 (2H, s), 4.63 (2H, s), 4.36-4.28 (1H, m), 3.78 (1H, dd, J = 3.4, 10.0 Hz), 3.74 (1H, dd, J = 5.9, 10.0 Hz), 3.35 (3H, s), 3.04 (1H, dd, J = 6.6, 13.0 Hz), 2.97 (1H, dd, J = 4.9, 13.0 Hz), 2.62 (3H, s); ¹³C NMR (CDCl₃, 100MHz) 155.7, 136.3, 128.4, 128.2, 128.0, 96.6, 68.4, 66.6, 56.1, 55.3, 47.5, 39.1; IR (neat) 3326 (m), 2928 (m), 1726 (m), 1687(s), 1541(s), 1466 (w), 1309 (m), 1273 (m),1142(m), 1039 (m), 1026(m), 727 (w); FABMS (m/z) 318 (10), 317 (22), 316 (MH+, 100), 284 (8), 176 (9), 164 (6), 91 (100); Exact MS (m/z) calcd. for C₁₄H₂₂O₅NS, 316.1218; found, 316.1202.
- 15. Under the stereoselective oxidation conditions for 10 to 11, the reaction did not proceed.
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- 17. Synthetic materials were spectroscopically (IR, ¹H NMR, ¹³C NMR, MS) identical with natural sparoxomycin A₁ and A₂. ^{4a}
- 18. The morphological reversion activity on sπ^{ts}NRK cells was assessed with HPLC pure compounds as follows; The cells were cultured in EAGLE's minimal essential medium (MEM) supplement with 10 % calf serum (CS, Hyclone Laboratories, Logan, Utah) at permissive temperature (32°C) or at nonpermissive temperature (39°C). The cells (1x10⁵ cells/ml) maintained at 32°C were seeded into a 96-well microtiter plate and cultured for two hours at 32°C in 5 % CO₂ atmosphere. Solution of various concentration of the compounds (5 μl each) was added and morphological reversion of sπ^{ts}NRK cells were observed under a microscope after 18 to 20 hours incubation at 32°C.

a minimal effective concentration. b no activity at 530µM